# Chemistry 510 Special Topics in Inorganic Chemistry

Professor J. Kovacs Wed & Fri 9:00–10:20 am Bagley 331A

#### Concepts:

1. Evidence of M<sup>+n</sup> importance to life

[M<sup>+n</sup>] in ocean vs body

Number of Metalloenzyme structures in Protein data bank

Number of Gordon conferences devoted to the topic

Workshops

2. Key Roles of M<sup>+n</sup> in Biology

Why transition-metals?

3. Metal ion properties important for catalytic activity

Labile M-L bonds

Ligand Field and Covalent bonds

Spin-state and Magnetic properties

Metal ion Lewis acidity

Redox potentials

4. Selecting the right M<sup>+n</sup> for a given function

Chelate effect/macrocyclic effect

M<sup>+n</sup> radius

5. How Nature tunes M<sup>+n</sup> properties

H-bonding

**Entatic State** 

Protein constraints

6. How M<sup>+n</sup> can be spectroscopically probed

Overview

7. Biological O<sub>2</sub> Chemistry

Overview

**Kinetics** 

Spin-Forbidden

Thermodynamics

Proton-coupled electron transfer (PCET)

8. Thermodynamics and kinetics of electron transfer

Redox potentials

Marcus theory

Role of protons in making e- transfer energetically feasible

9. The electron transport chain

Why are Fe<sub>4</sub>S<sub>4</sub> clusters and Blue Cu so good at electron transfer?

Mossbauer

**EPR** 

## 10. Controlled Dioxygen Activation

Mechanism overview

Assessing extent of activation using rRaman

Superoxo intermediates in P450 and IPNS

## 11. H-atom Transfer and PCET in Biological Oxidation Rxns

Are Fe-O<sub>2</sub>\*- potent enough to cleave C-H bonds?

Deuterium isotope effects

Hydroperoxo intermediates (homolytic and heterolytic cleavage)

Binuclear peroxo intermediates

High-valent Fe(IV)=O and Fe(V)=O intermediates

The oxo wall

Electronic structure of high-valent oxos

Thermodynamic cycle of HAT

Asynchronous PCET

### 12. Electronic structure Fe(IV)=O compounds

Spin-state dependent reactivity

Incipient formation of Fe-oxyl

#### 13. Nitrogenase

14. Photosynthetic water oxidation